## A Good GC Stationary Phase for Separation of Xylene Isomers

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**Abstract:** An excellent separation of xylene isomers is reported using 2, 6-O-di-pentyl-3-O-acetyl- $\beta$ -CD as capillary stationary phase. The optimum conditions for the separation are also discussed.

**Keywords:** Xylene isomers separation; capillary stationary phase; optimum conditions for separation.

The separation of xylene isomers using GC is a much difficult task for a long time, on the other hand they are very important industrial materials, so finding a good stationary phase has been an interesting work. An excellent separation of xylene isomers was obtained by using 2, 6-di-O-pentyl-3-O-acetyl- $\beta$ -cyclodextrin synthesized in our lab as the capillary stationary phase. The chromatograms and separation factor ( $\alpha$ ) are better than those reported in the literature. We hope this will provide a good method for the separation of xylene isomers for practical use.

Preparation of Heptatakis (2, 6-di-O-pentyl-3-O-acetyl- $\beta$ -cyclodextrin):  $\beta$ -cyclodextrin was subjected to pentylation with sodium hydroxide and n-pentyl bromide in DMSO according to the method of Ciucanu and Kerek<sup>1</sup>. The resulted heptakis (2, 6-di-O-pentyl) - $\beta$ -cyclodextrin was further acetylated with acetic anhydride and dry pyridine by heating the solution at 80 °C for 3 hours. After removing the solvent in vacuum the residue was purified by column chromatography on silica gel.

**Table I** gives the chromatographic data at different conditions.

**Table I** shows that in the temperature range of 90 °C to 110 °C changes of temperature and carrier gas velocity have little effect on the separation factor ( $\alpha$ ). But when the temperature was out of this range, either lower  $\alpha$  values or worse peak shapes appeared. It indicates that the optimum temperature range for xylene isomers separation is 90-110 °C.

Other stationary phases listed in **Table II** either give smaller  $\alpha$  values than those in **Table I** or could not separate three xylene isomers completely.

Temp. Velocity of carrier Temp. Velocity of carrier gas (°C) gas (cm/s) (°C) (cm/s) 5.22 1.52 90 17.86 6.07 1.16 110 18.82 1.16 2.82 6.91 1.14 3.19 1.13 6.06 2.30 16.20 7.01 1.16 15.77 2.67 1.16 7.99 1.14 3.04 1.14 2.15 3.24 m m 100 14.23 3.76 1.16 17.16 2.46 1.15 p 4.28 1.14 2.75 1.12 2.97 2.61 m 1.16 m 19.23 22.03 3.46 1.14 p 2.96 1.13 3.95 3.32 1.12

**Table I.** Capacity factor (k) and separate factor  $(\alpha)$  for xylem isomers

• The values listed are for two adjacent eluting isomers.

Table II. Related data reported in literature

Ref.	Stationary phase	Temp.	k	α	Ref.	Stationary phase	Temp.	k	α
		(℃)					$(^{\circ}C)$		
	Crown-ether		m 2.00			Peralkylated β -CD		p 1.517	
4	bonded β -CD	80	p 2.00	1.00	3		80	m 1.558	1.03
			o 2.67	1.33				o 1.817	1.17
			m 2.46					m /	
4	Resorararenes <sup>4</sup>	80	p 2.46	1.00	2	Heptakis (2, 6-O-	60	p /	1.00
			o 3.17	1.29		di-benzyl-3-O-		o /	1.19
						acetyl) - β -CD			
	Metal complex		m 0.80						
4	liquid crystalline*	45	p 1.33	1.66					
			o 1.44	1.19					

<sup>\*</sup>A baseline separation for m- and p- xylene isomers was not obtained.

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